Tetrahedron Letters 50 (2009) 3414-3417

Contents lists available at ScienceDirect

**Tetrahedron Letters** 

journal homepage: www.elsevier.com/locate/tetlet

# Cyclopropane ring-opening of tricyclo[3.3.0.0<sup>2,8</sup>]octan-3-ones: a quick access to bicyclo[3.2.1]octanones from 2-methoxyphenols

Chun-Ping Chang<sup>a</sup>, Ching-Hsien Chen<sup>a</sup>, Gary Jing Chuang<sup>a</sup>, Chun-Chen Liao<sup>a,b,\*</sup>

<sup>a</sup> Department of Chemistry, National Tsing Hua University, Hsinchu 30013, Taiwan
<sup>b</sup> Department of Chemistry, Chung Yuan Christian University, Chungli 32023, Taiwan

#### ARTICLE INFO

ABSTRACT

Article history: Received 13 January 2009 Revised 16 February 2009 Accepted 19 February 2009 Available online 25 February 2009 We herein report an efficient and rapid strategy for the synthesis of highly functionalized bicyclo[3.2.1]octanones via sequential reactions: Diels-Alder reaction of MOB-ODPM rearrangementreductive cleavage of cyclopropane, and its application to synthesize the core structure of drechslerine D. © 2009 Elsevier Ltd. All rights reserved.

Bicyclic structures have been one of the main spotlights in synthetic chemistry over the years owing to their presence in a large number of natural products and synthetic challenges.<sup>1</sup> Among the derivatives of bicyclo[3.2.1] skeleton, drechslerine,<sup>2</sup> a small and rare family of sesquiterpenoids (Fig. 1), has drawn our attention with its structural uniqueness. Although there have been quite a few synthetic approaches developed over the years for the construction of bicyclo[3.2.1] skeletons through diverse chemical methodologies,<sup>3</sup> synthesis of the drechslerine family natural products is yet to be reported. The installations of the fused lactone ring and the quaternary bridgehead on this bicyclic structure post a great challenge for synthetic chemists.

In the past, we have successfully demonstrated the power and potential of the chemistry of masked o-benzoquinones (MOBs) in organic syntheses to construct highly functionalized bicyclo[2.2.2]octenones,<sup>4</sup> cis-decalins,<sup>5</sup> bicyclo[4.2.2]- and bicyclo[3.2.2]-skeletons,<sup>6</sup> linear and angular triguinanes<sup>7</sup> skeletons from the Diels-Alder adducts of MOBs, and applied these synthetic strategies to the syntheses of natural products with diverse structural nature.<sup>8</sup> The strategy that we have employed for the syntheses of triquinane/diquinane cores of capnellene,<sup>8d</sup> magellanine,<sup>9</sup> and forsythide aglucone dimethyl ester<sup>10</sup> is the oxa-di- $\pi$ -methane (ODPM) rearrangements of bicyclo[2.2.2]octenones, adducts from the Diels-Alder reactions of MOBs, followed by the reductive cleavage of the tricyclo[3.3.0.0<sup>2,8</sup>]octan-3-ones (Scheme 1). As the cleavage of 'a-bond' yielded the diquinane skeleton, we foresee the cleavage of '**b**-bond' of the ODPM rearrangement products would give an access to the corresponding bicyclo[3.2.1] skeletons. Though it has been reported that the cleavage of the cyclopropane ring in this tricyclic system typically took place at '**a**-bond',<sup>7,11</sup> examples of using nucleophilic addition<sup>12</sup> or functional group directing reductive cleavage<sup>13</sup> to generate bicyclo[3.2.1]octanones from compounds of tricyclo[3.3.0.0<sup>2,8</sup>] skeleton have also been reported. Thus as the continuation of our efforts to develop new and efficient methodologies for the syntheses of natural products starting from 2-methoxy-





OMe

bicyclo[3.2.1]octan-3-ones

OMe

Ò





<sup>\*</sup> Corresponding author. Tel.: +886 3 5715131 33388; fax: +888 3 5728123. *E-mail addresses*: ccliao@mx.nthu.edu.tw, ccliao@cycu.edu.tw (C.-C. Liao).

<sup>0040-4039/\$ -</sup> see front matter  $\odot$  2009 Elsevier Ltd. All rights reserved. doi:10.1016/j.tetlet.2009.02.150

phenols via the corresponding MOBs, we report herein our results of the constructions of bicyclo[3.2.1]octanones from 2-methoxyphenols via sequential reactions: *Diels–Alder reaction of MOB–ODPM rearrangement–reductive cleavage of cyclopropane.* 

In order to induce the desire selectivity of cleaving '**b**-bond' over '**a**-bond' in the forthcoming reductions of tricyclo[ $3.3.0.0^{2.8}$ ]octan-3-ones obtained from the photoreaction,<sup>13</sup> we deliberated to utilize MOBs with electron-withdrawing groups (R = CO<sub>2</sub>Me, COMe, or CN), thus making the **b**-bond, connecting to the keto group of cyclopentanone and the R-substituent on the cyclopropane ring to be especially electron deficient and prone for the reduction to take place. The installation of the electron-withdrawing R-groups could also avoid the possibility of unwanted decarbonylation or other types of rearrangements during the triplet sensitized photoreactions.<sup>14</sup> The results of the Diels–Alder reactions of selected MOBs<sup>8c,8d</sup> and the following irradiation with 300 nm UV light in acetone with acetophenone as the sensitizer in a Rayonet reactor to give the ODPM rearrangement products are shown in Table 1.

The cycloadditions of MOBs and the subsequent photorearrangements gave moderate to excellent yields to the tricyclo[3.3.0.0<sup>2,8</sup>]octan-3-ones, precursors for the next stage cyclopropane ring-opening reaction. The detailed experimental procedures and structural determinations with spectral data assignment are available in the Supplementary data. After having compounds 5 and 6 in hand, we tried a number of reducing conditions. Messy mixtures were obtained from reactions using Li/NH<sub>3</sub><sup>15</sup> and SmI<sub>2</sub><sup>7a,16</sup> as reducing agents. Treatments of **5** and **6** with Zn/AcOH were found unreactive.<sup>17</sup> However, reduction using *n*-Bu<sub>3</sub>SnH, reported by Enholm, <sup>13a,13b</sup> Lahiri, <sup>13c</sup> and us, <sup>7b,9</sup> was found to give the ring-opening reaction in clean and good yields (Table 2). The reductive cleavage reactions were directed by the adjacent carbonyl and electron-withdrawing R-groups and found to be completely chemoselective for the cleavage of '**b**-bond'. The desired bicyclo[3.2.1] octanones were obtained in moderate to good yields. As we examined closely the stereochemistry of the reduced products, an inversion of configuration at the carbon center of the cyclopropane where the R-group is attached was found during the bond-breaking process (Fig. 2). This stereo-chemical outcome is presumably due to steric effect. In which during the cyclopropane ring-opening reaction, the late stage hydrogen abstraction of inter-

#### Table 1

Construction of tricyclo[3.3.0.0<sup>2,8</sup>]octan-3-ones from MOBs

	$R = CO_2Me$ <b>1a</b> : R = CO_2Me <b>1b</b> : R = COMe <b>1c</b> : R = CN	R OMe OMe MOB 2	$\begin{array}{c} X \\ OMe \\ OMe \\ OMe \\ CMe \\ CMe \\ R \\ X = COMe, 3 \\ X = Ph, 4 \end{array} \qquad \begin{array}{c} OMe \\ OMe \\ R \\ X = COMe, 5 \\ X = Ph, 6 \end{array}$	
Entry	MOB	Х	D-A adduct (%)	ODPM product (%)
1	2a	COMe	<b>3a</b> /87	<b>5a</b> /83
2	2b	COMe	<b>3b</b> /85	<b>5b</b> /93
3	2c	COMe	<b>3c</b> /44	<b>5c</b> /86
4	2a	Ph	<b>4a</b> /95	<b>6a</b> /90
5	2b	Ph	<b>4b</b> /93	<b>6b</b> /95
6	2c	Ph	<b>4c</b> /90	<b>6c</b> /96

<sup>a</sup>Diacetoxyiodobenzene. <sup>b</sup>Photolysis condition: hv = 300 nm, 10 equiv acetophenone, substrate as 1% solution in acetone.

#### Table 2

Reductive cleavage to yield bicyclo[3.2.1]octanones



Entry	R	Х	Product (%)
1 ( <b>5a</b> )	CO <sub>2</sub> Me	COMe	<b>7a</b> /59
2 ( <b>5b</b> )	COMe	COMe	<b>7b</b> /64
3 ( <b>5c</b> )	CN	COMe	<b>7c</b> /51
4 ( <b>6a</b> )	CO <sub>2</sub> Me	Ph	<b>8a</b> /65
5 ( <b>6b</b> )	COMe	Ph	<b>8b</b> /83
6 ( <b>6c</b> )	CN	Ph	<b>8c</b> /40



Figure 2. ORTEP drawing for 7b at 50% probability thermal ellipsoids.

mediate **II** took place from the less hindered side of the tertiary radical center to avoid steric interaction to the dimethoxy ketal and bulky enol connected SnBu<sub>3</sub> group as shown in Table 2's drawing.

After successfully obtaining bicyclo[3.2.1]octanones from the *Diels–Alder reaction of MOB–ODPM rearrangement–reductive cleav-age of cyclopropane* sequence, we then applied this methodology to the synthesis of the core structure of drechslerine D (Scheme 2). As a contrast to high yielding Diels–Alder reactions mentioned in Table 1 and in our previous results,<sup>8c</sup> initial Diels–Alder reaction of MOB **2d** with furanone **9** was found sluggish under conventional heating method. However, using microwave reactor significantly shortens the reaction time and gave a moderate yield of the cycloadducts. Unfortunately, a 1: 1 ratio of regioselectivity was found from the cycloaddition to give isomeric compounds **3d** and **3d'**. Carrying the desired regiochemistry, adduct **3d'** was hydrolyzed to give aldehyde **10**, which was then irradiated accordingly to the previously mentioned condition to produce compound **5d** in good yield. The following reductive cleavage with *n*-Bu<sub>3</sub>SnH furnished



Figure 3. ORTEP drawing for 7d at 50% probability thermal ellipsoids.

**7d.** Thus within only four chemical transformations, a bicyclo[3.2.1]octanone with 3 (the fused carbons on lactone ring and the quaternary carbon on the bridgehead of the bicyclo[3.2.1]octanone) of the 4 stereocenters of drechslerine D was synthesized (Fig. 3).

In summary, we have successfully developed an efficient methodology for the syntheses of highly functionalized bicyclo [3.2.1]octanone systems. Several bicyclo[3.2.1]octanone derivatives were obtained rapidly in good to excellent yields starting from commercially available or easily accessible 2-methoxyphenols. In addition, the efficacy of our strategy has been demonstrated by utilizing it as a key step for the synthesis drechslerine D framework.

## Acknowledgments

We gratefully acknowledge the financial supports from National Science Council of Taiwan (NSCT). C.C.L. thanks Chung Yuan Christian University for the support under Grant CYCU-97-CR-CH and G.J.C. thanks NSCT for a postdoctoral fellowship.



Scheme 2. Synthesis of the core structure of drechslerine.

### Supplementary data

Crystallographic data of **3d**', **5c**, **7b**, and **7d** have been deposited with the Cambridge Crystallographic Data Centre as supplementary publication numbers: CCDC 715844, 715845, 715846, 715847, respectively. Copies of the data can be obtained, free of charge, on application to CCDC, 12 Union Road, Cambridge CB2 1EZ, UK, (fax: +44-(0)1223-336033 or e-mail: deposit@ccdc.cam.ac.Uk). Supplementary data associated with this article can be found, in the online version, at doi:10.1016/j.tetlet.2009. 02.150.

# **References and notes**

- Pirrung, M. C.; Morehead, A. T., Jr.; Young, B. G. A sesquidecade of sesquiterpenes: total synthesis, 1980–1994. Part B: bicyclic and tricyclic sesquiterpenes. In *The Total Synthesis of Natural Products*; Goldsmith, D., Ed.; John Wiley & Sons: New York, 2000; Vol. 11, pp 186–199, and references therein.
- (a) Osterhage, C.; König, G. M.; Höller, U.; Wright, A. D. J. Nat. Prod. 2002, 65, 306; (b) Fraga, B. M. Nat. Prod. Rep. 2003, 20, 392.
- (a) Rodriguez, J.; Filippini, M. H. Chem. Rev. **1999**, 99, 27; (b) Danishefsky, S.; Vaughan, K.; Gadwood, R.; Tsuzuki, K. J. Am. Chem. Soc. **1981**, 103, 4136; (c) Paquette, L. A.; Reagan, J.; Schreiber, S. L.; Teleha, C. A. J. Am. Chem. Soc. **1989**, 111, 2331; (d) Corey, E. J.; Liu, K. J. Am. Chem. Soc. **1997**, 119, 9929; (e) Iguchi, K.; Ito, H.; Takeguchi, S.; Kawagishi, T. Org. Lett. **2006**, 8, 4883.

- 4. Chu, C.-S.; Lee, T.-H.; Liao, C.-C. Synlett 1994, 635.
- (a) Lee, T.-H.; Liao, C.-C.; Liu, W.-C. Tetrahedron Lett. **1996**, 37, 6819; (b) Hsiu, P.-Y.; Liao, C.-C. Chem. Commun. **1997**, 1086; (c) Rao, P. D.; Chen, C.-H.; Liao, C.-C. Chem. Commun. **1998**, 155; (d) Hsu, P.-Y.; Lee, Y.-C.; Liao, C.-C. Tetrahedron Lett. **1998**, 39, 659.
- Lu, Y.-B.; Lee, T.-H.; Liu, W.-C.; Chuang, G. J.; Liao, C.-C. Chem. Asian J. 2008, 3, 1422.
- (a) Hwang, J.-T.; Liao, C.-C. Tetrahedron Lett. 1991, 32, 6583; (b) Hsu, D.-S.; Rao, P. D.; Liao, C.-C. Chem. Commun 1998, 1795.
- (a) Chu, C.-S.; Liao, C.-C.; Rao, P. D. Chem. Commun. **1997**, 1537; (b) Lee, T.-H.; Liao, C.-C. Tetrahedron Lett. **1996**, 37, 6869; (c) Liao, C.-C.; Peddinti, R. K. Acc. Chem. Res. **2002**, 35, 856; (d) Liao, C.-C. Pure. Appl. Chem. **2005**, 77, 1221; (e) Hsu, D.-S.; Hsu, P.-Y.; Lee, Y.-C.; Liao, C.-C. J. Org. Chem. **2008**, 73, 2554.
- 9. Yen, C.-F.; Liao, C.-C. Angew. Chem., Int. Ed. 2002, 41, 4090.
- 10. Wei, C.-P.; Liao, C.-C. Tetrahedron Lett. 1989, 30, 2255.
- 11. Yates, P.; Stevens, K. E. Tetrahedron 1981, 37, 4401.
- 12. Imanishi, T.; Matsui, M.; Yamashita, M.; Iwata, C. Tetrahedron Lett. **1986**, 27, 3161.
- (a) Enholm, E. J.; Jia, Z. J. *Tetrahedron Lett.* **1995**, *36*, 6819; (b) Enholm, E. J.; Jia, Z. J. J. Org. Chem. **1997**, *62*, 174; (c) Yadav, S.; Banerjee, S.; Maji, D.; Lahiri, S. *Tetrahedron* **2007**, *63*, 10979.
- (a) Katayama, S.; Yamauchi, M. Chem. Lett. 1995, 311; (b) Katayama, S.; Hiramatsu, H.; Aoe, K.; Yamauchi, M. J. Chem. Soc., Perkin Trans. 1 1997, 561; (c) Lee, T.-H.; Rao, P. D.; Liao, C.-C. Chem. Commun. 1999, 801.
- 15. Castellino, A. J.; Bruice, T. C. J. Am. Chem. Soc. 1988, 110, 7512.
- 16. Kim, Y.-H.; Lee, I.-S. Heteroat. Chem. 1992, 3, 509.
- (a) Murphy, W. S.; Wattanasin, S. J. Chem. Soc., Perkin Trans. 1 1986, 8, 1445; (b) Barton, D. H. R.; Gilham, P. T. J. Chem. Soc. 1960, 4596.